# **IRIDOIDS FROM GALIUM MOLLUGO\***

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Abstract—From Galium mollugo, two new iridoids, gardenosidic acid and 10-hydroxymorroniside as well as 10-hydroxyloganin have been isolated, along with seven known iridoids, secogalioside, asperuloside, asperulosidic acid, daphylloside, monotropein, scandoside and scandoside methyl ester 10-Hydroxyloganin, a compound which was previously considered to be the key biosynthetic intermediate of secoiridoids, was obtained for the first time from a natural source

#### INTRODUCTION

It has been reported that Galium mollugo L. (G album Mill) contains the following iridoid glucosides asperuloside (1) [1, 2], asperulosidic acid (2), monotropein (3), galioside (monotropein methyl ester) (4) [2] and mollugoside (5) [3], as well as a secoiridoid glucoside secogalioside (6) [1] No other instance has so far been recorded of the co-occurrence of a secoiridoid glucoside with iridoid glucosides possessing a highly oxidized cyclopentan ring The present paper deals with a reexamination of the iridoid constituents of G mollugo plants §

### RESULTS AND DISCUSSION

The methanolic extract of the fresh aerial parts of G mollugo was fractionated by a combination of charcoal CC, silica gel CC and prep TLC As a result, two new iridoids, gardenosidic acid (7) and 10-hydroxymorroniside (8), as well as 10-hydroxyloganin (9) were isolated The latter glucoside (9), which had already been chemically prepared [4-6] was obtained for the first time from a natural source Additionally, three known iridoids scandoside (10), scandoside methyl ester (11) and daphylloside (12) were newly isolated from this plant, besides the reported iridoids asperuloside (1), asperulosidic acid (2), monotropein (3) and secogalioside (6)

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§ In the present work, we were unable to detect galioside (4) and mollugoside (5) This might be due to aerial or seasonal variations in the metabolism of the plant

|| As the <sup>1</sup>H NMR spectrum of the fraction containing 7 did not show any signals from methoxy or acetyl groups, and that of the mixture of 8 and 9 showed no signals from acetyl groups, 7 was purified as its methyl ester acetate, and 8 and 9 were separated as their acetates

Gardenosidic acid (7)

This substance was purified as its methyl ester acetate, a white powder,  $C_{29}H_{36}O_{17}$ ,  $[\alpha]_{D}^{18} - 70.7^{\circ}$  (MeOH, c 1.15), which was identical in all respects with an authentic specimen of gardenoside hexaacetate (13) [7, 8]

10-Hydroxyloganın (9)|| and 10-hydroxymorroniside (8)||

A mixture of these two glucosides was separated after acetylation to give 10-hydroxyloganin hexaacetate (14), (7R)-10-hydroxymorroniside hexaacetate (15) and (7S)-10-hydroxymorroniside hexaacetate (16)

10-Hydroxyloganin hexaacetate (14), a white powder,  $C_{29}H_{38}O_{17}$   $H_2O$ ,  $[\alpha]_{2}^{D2}$  – 52 4° (CHCl<sub>3</sub>, c 0 42), showed spectral data (see below) in accord with those of 10-hydroxyloganin hexaacetate which was chemically prepared [4–6] UV  $\lambda_{\max}^{ECH}$  nm (log  $\varepsilon$ ) 232 (4 02), IR  $\nu_{\max}^{KBR}$  cm<sup>-1</sup> 1755, 1740, 1705 (sh), 1640, <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ 1 92–2 09 (6 × OAc), 3 06 (ddd, J = 14 0, 8 0, 1 5 Hz, H-5), 3 70 (s, COOMe), 4 06–4 36 (m, 10-H<sub>2</sub>, 6'-H<sub>2</sub>), 5 26 (d, J = 3 0 Hz, H-1), 7 33 (d, J = 1 2 Hz, H-3) The Zemplén reaction of the acetate 14 gave a powdery free glucoside,  $[\alpha]_{2}^{D2}$  – 58 9° (MeOH, c 0 89), which was identical in all respects with synthetic 10-hydroxyloganin (9) [4–6] Compound 9 was originally considered to be the key biosynthetic intermediate between loganin and secologanin [9], but this possibility was recently ruled out [10, 11]

(7R)-10-Hydroxymorroniside hexaacetate (15) and (7S)-10-hydroxymorroniside hexaacetate (16), each  $C_{29}H_{38}O_{18}$ , were obtained in a 5 1 ratio The physical properties of 15 and 16 (shown below) suggested that 15 and 16 were closely related to (7R)-morroniside pentaacetate (17) and (7S)-morroniside pentaacetate (18), respectively (7R)-Isomer (15), a white powder,  $[\alpha]_{12}^{D}$  -68 1° (CHCl<sub>3</sub>, c 1 09), UV  $\lambda_{max}^{EnoH}$  nm (log ε) 235 (4 09), IR  $\nu_{max}^{RBT}$  cm<sup>-1</sup> 1730, 1700, 1630, <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ2.01–2 10 (6 × OAc), 2 90 (dt, J = 13 0, 5 0 Hz, H-5), 3 73 (s, COOMe), 4 00 (m, H-8), 4 08–4 40 (m, 10-H<sub>2</sub>, 6'-H<sub>2</sub>), 5 66 (d, J = 9 0 Hz, H-1), 5 74 (dd, J = 10 0, 2 5 Hz, H-7), 7 44 (s, H-3) (7S)-Isomer (16), colourless needles, mp 183 5°,  $[\alpha]_{D}^{27}$  – 101 3° (CHCl<sub>3</sub>, c 1 00), UV  $\lambda_{mon}^{EIOH}$  nm

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(log  $\varepsilon$ ) 235 (402), IR  $v_{\text{max}}^{\text{KBr}}$  cm<sup>-1</sup> 1710, 1680, 1615, <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ 200–212 (6 × OAc), 312 (dt, J = 12 5, 5 0 Hz, H-5), 3 73 (s, COOMe), 4 10-4 42 (m, H-8,  $10-H_2$ , 6'- $H_2$ ), 5 66 (d, J = 90 Hz, 1-H), 6 16 (d (br), J= 30 Hz, H-7), 7 44 (s, H-3) In keeping with the above suggestion, the <sup>13</sup>C NMR spectra (Table 1) of 15 and 16 were in good accord with those of 17 and 18, respectively, except that the 10-methyl signals (17  $\delta$ 18 8, 18  $\delta$ 18 8) were replaced by 10-hydroxymethyl signals (15  $\delta$ 64 6, 16  $\delta$ 64 8) Accordingly, it was presumed that 15 was the 10acetoxymethyl congener of (7R)-morroniside pentaacetate (17), whereas 16, that of (7S)-morroniside pentaacetate (18) This presumption received support from the following <sup>1</sup>H and <sup>13</sup>C NMR analyses in the <sup>1</sup>H NMR spectrum of 15, strong NOEs were observed between H-5, H-7 and H-8, indicating that these three protons in the chair tetrahydropyran ring of 15 were in a 1,3,5-triaxial relationship Thus, the chirality at both C-7 and C-8 of 15 was proved to be R Furthermore, in the 13C NMR spectra, the C-5 and C-8 of 16 resonated 41 and 64 ppm upfield relative to the corresponding carbons of 15 Such discrepancy of the carbon chemical shifts was also observed between the C-5 and C-8 signals of 18 and 17 The observed upfield shift of C-5 and C-8 was explained in terms of reciprocal y effects of the C-7 acetoxy group on these carbons in the tetrahydropyran ring of 16 and 18 \*

Zemplén reaction of the major product 15 yielded 10-hydroxymorroniside (8),  $C_{17}H_{26}O_{12}$   $2H_2O$ ,  $[\alpha]_D^{22}-97.8^{\circ}$  (MeOH, c 0.54) as a white powder The accompanying spectral data of 8 clearly indicated that 8 exists, at least in methanol solution, as a mixture of (7R)-and (7S)-isomers in a 2.1 ratio in the same way as morroniside (19) [12] UV  $\lambda_{\rm max}^{\rm McOH}$  nm (log  $\varepsilon$ ) 238 (4.19), IR  $\nu_{\rm max}^{\rm KBr}$  cm<sup>-1</sup> 3350, 1700 (sh), 1690, 1646 (sh), 1625, <sup>1</sup>H NMR (CD<sub>3</sub>OD) (7R)-isomer,  $\delta$ 1 23 (m, H-6<sub>ax</sub>), 193

Table 1 <sup>13</sup>C NMR data\* of compounds 15-18 (50 10 MHz, CDCl<sub>3</sub>, TMS as int standard)

c	15	16	17	18	
1	94 5 d	94 0 d	94 8 d	94 4 d	
3	1524d	152 4 d	152 5 d	152 5 d	
4	109 7 s	1104s	1103s	111 1 s	
5	304d	26 3 d	30 1 d	26 0 d	
6	33 5 t	31 8 t	33 1 ε	31 3 t	
7	940d	90 9 d	93 9 d	91 3 d	
8	758d	69 4 d	73 6 d	67 3 d	
9	36 5 d	36 6 d	38 9 d	39 2d	
10	64 6 t	64 8 t	188q	18 8 q	
11	166 3 s	166 4 s	166 4 s	166 6 s	
OMe	51 5 q	51 4 q	51 3 q	51 3 q	
1'	974d	97 2 d	968d	96 8 d	
2'	71 0 d	71 0 d	71 0 d	71 1 d	
3′	72 2 d	72 2 d	72 1 d	72 1 d	
4′	68 8 d	68 4 d	68 6 d	687 d	
5′	726d	72 5 d	72 6 d	72 6 d	
6′	61 8 t	61 5 t	61 8 t	617t	

\*Off resonance patterns are given after the chemical shift value Each compound has additional signals arising from acetoxy groups

(m, H-9), 2 10  $(ddd, J = 135, 40, 20 \text{ Hz}, \text{H-6}_{eq})$ , 2 85 (dt, J = 125, 50 Hz, H-5), 3 70 (s, COOMe), 5 78 (d, J = 95 Hz, H-1), 7 50 (s, H-3) (7S)-Isomer,  $\delta$ 1 54  $(dt, J = 135, 35 \text{ Hz}, \text{H-6}_{ax})$ , 1 93  $(m, \text{H-9}, \text{H-6}_{eq})$ , 3 70 (s, COOMe), 4 45 (m, H-8), 5 28 (d, J = 35 Hz, H-7), 5 84 (d, J = 95 Hz, H-1), 7 50 (s, H-3)

### **EXPERIMENTAL**

General procedure Mps uncorr, <sup>1</sup>H NMR (200 MHz) and <sup>13</sup>C NMR (50 10 MHz) CDCl<sub>3</sub> or CD<sub>3</sub>OD with TMS as int standard, TLC silica gel 60 GF<sub>254</sub>, spots visualized by irradiation with UV light (254 nm), exposure to I<sub>2</sub> vapour or spraying with anisaldehyde–H<sub>2</sub>SO<sub>4</sub> reagent (anisaldehyde 0.5 ml, 95% EtOH 9.0 ml and AcOH 0.1 ml) followed by heating, Prep TLC silica gel 60 PF<sub>254</sub>, bands detected by irradiation with UV light (254 nm) The bands due to free glucosides were extracted with CHCl<sub>3</sub>–MeOH (7.3), while the ones due to the acetates were extracted with CHCl<sub>3</sub>–MeOH (19.1), CC carbon (Wako and Takeda) or silica gel PF<sub>254</sub>

Plant material The seeds of Galium mollugo were donated by Drs S R Jensen and B J Nielsen (Technical University of Denmark) They were planted at the Medicinal Plant Garden, Faculty of Pharmaceutical Sciences, Kyoto University in October 1980, and the plants were collected in July 1981 A voucher specimen of G mollugo (H Inouye No 1) has been deposited in the Herbarium of the Institute of Botany, Faculty of

Fig 1 Conformation of compounds 15 and 16

<sup>\*</sup>This finding suggests that the tetrahydropyran ring of both compounds also assumes a chair conformation, although we previously surmised that the tetrahydropyran ring of 16 is in a boat form [12]

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Isolation of iridoids Fresh aerial parts (119 kg) were extracted with boiling MeOH (501 × 3) After concentration of the combined extracts in vacuo at 40°, the residue was diluted with H<sub>2</sub>O (151), and the insoluble material was filtered off through a Celite layer, which was then washed with H<sub>2</sub>O (101) The combined filtrate and washings were concentrated in vacuo to give a residue (3445g) which was chromatographed on a charcoal (14 kg) column developed with EtOH-H2O mixtures containing increasing amounts of EtOH This gave the following fractions Fr 1, the faster 10% EtOH (725g) Fr 2, slower 10% EtOH (79 g) Fr 3, 40% EtOH (80 g) Fr 4, 80% EtOH (30 g) Fr 1 was extracted with boiling EtOH (100 ml × 3) and the residue (17g) of the EtOH extract treated with excess ethereal CH<sub>2</sub>N<sub>2</sub> in MeOH at 0° for 15 min. The resultant ppt was filtered off and washed with MeOH (10 ml × 3) The combined filtrate and washings were concentrated in vacuo to give a residue (105 g), which was subjected to silica gel (30 g) CC with CHCl<sub>3</sub>-MeOH of an increasing MeOH content The residue (184 mg) of the 10% MeOH-CHCl<sub>3</sub> eluate was subjected to prep TLC with CHCl3-MeOH (4 1, developed × 3) The less mobile major band afforded crude glucoside (46 mg) This was acetylated with pyridine-Ac2O (each 05 ml) at 30° for 15 hr and the product (60 mg) purified by prep TLC (four developments) with  $C_6H_6$ -Et<sub>2</sub>O (3 1) to yield gardenoside hexaacetate (13) (16 mg) as a white powder An aliquot (257 g) of Fr 2 was rechromatographed on a charcoal (15 g) column developed successively with  $H_2O(11)$ , 7.5% MeOH(11) and MeOH(11) The MeOH eluate afforded a mixture (180 mg) of monotropein (3) and scandoside (10), which was successively methylated and acetylated in the conventional way. The product was then subjected to prep TLC (CHCl3-MeOH, 20 1) Of two major bands, the more mobile one gave the hexaacetate (130 mg) of scandoside methyl ester (11) as colourless needles, mp 133-135°, whereas the less mobile band furnished the pentaacetate (18 mg) of monotropein methyl ester (4) as colourless needles, mp 148-153° An aliquot (5 g) of Fr 3 was chromatographed on a silica gel (120 g) column with CHCl3-MeOH of an increasing MeOH content The 5% MeOH-CHCl3 eluate gave first crystalline asperuloside (1, 1600 mg) and secondly powdery secogalioside (6, 111 mg) The 75% MeOH-CHCl<sub>3</sub> eluate furnished a residue (47 mg), which was subjected to prep TLC (CHCl<sub>3</sub>-MeOH, 4 1, developed  $\times$  3) to afford scandoside methyl ester (11, 9 mg) as a white powder The 10% MeOH-CHCl<sub>3</sub> eluate gave a residue (1530 mg), an aliquot (471 mg) of which was submitted to prep TLC (CHCl3-MeOH, 4 1, developed × 4). Of two major bands, the more mobile one gave a glucoside fraction A (254 mg), whereas the less mobile one, a glucoside fraction B (110 mg) Fraction A was acetylated and the product (336 mg) subjected to prep TLC ( $C_6H_6$ - $Et_2O$ , 7 3, developed  $\times$  5) to give 10-hydroxyloganin hexaacetate (14, 113 mg), (7R)-10-hydroxymorroniside hexaacetate (15, 89 mg), and (7S)-10-hydroxymorroniside hexaacetate (16, 22 mg) in order of decreasing polarity 14, Found C, 51 49, H, 5 74 C<sub>29</sub>H<sub>39</sub>O<sub>17</sub> H<sub>2</sub>O requires C, 51 48, H, 596% 15, Found C, 51 79, H, 576 C<sub>29</sub>H<sub>38</sub>O<sub>18</sub> requires C, 51 62, H, 568% 16, Found C, 51 13, H, 581  $C_{29}H_{38}O_{18}$ requires C, 51 62, H, 5 68% Next, fraction B (110 mg) was subjected to prep TLC (CHCl $_3$ -MeOH, 4 1, developed  $\times$  4), and

the major band gave a white powder (78 mg), which was recrystallized from EtOH to afford asperulosidic acid (2, 55 mg) as colourless needles, mp 125–128° Fr 4 was chromatographed on a silica gel (60 g) column with CHCl<sub>3</sub>–MeOH of an increasing MeOH content The residue (388 mg) of the 5% MeOH–CHCl<sub>3</sub> eluate was subjected to prep TLC (CHCl<sub>3</sub>–MeOH, 4 1, developed × 3) The major band gave daphylloside (12, 207 mg) as a white powder

Zemplén reaction of 10-hydroxyloganin hexaacetate (14) Methanolic NaOMe (0 1 M, 0 1 ml) was added to a soln of 14 (50 mg) in MeOH (2 ml) and the mixture was refluxed for 10 min After cooling, the soln was neutralized with Amberlite IR-120B (H<sup>+</sup>-form) and concd in vacuo The residue (45 mg) was purified by prep TLC (CHCl<sub>3</sub>-MeOH, 4 1) to give 10-hydroxyloganin (9) as a white powder <sup>1</sup>H NMR (CD<sub>3</sub>OD)  $\delta$ 1 54 (ddd, J = 140, 100, 45 Hz, H-6<sub>ax</sub>), 208 (m, H-8, H-9), 226 (ddd, J = 140, 70, 15 Hz, H-6<sub>eq</sub>), 430 (deformed t, H-7), 516 (d, J = 60 Hz, H-1), 746 (d, J = 12 Hz, H-3) (Found C, 49 26, H, 645 C<sub>17</sub>H<sub>26</sub>O<sub>11</sub>  $\frac{1}{2}$  H<sub>2</sub>O requires C, 49 15, H, 655%)

Zemplén reaction of (7R)-10-hydroxymorroniside hexaacetate (15) Methanolic NaOMe (0 1 M, 0 1 ml) was added to a soln of 15 (107 mg) in MeOH (1 ml) and the mixture was allowed to stand for 30 min in iced  $\rm H_2O$  The soln was worked up in the same way as above to afford 10-hydroxymorroniside (8, 52 1 mg) as a white powder (Found C, 44 71, H, 6 58  $\rm C_{17}H_{26}O_{12}$  2 $\rm H_2O$  requires C, 44 54, H, 6 60%)

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# REFERENCES

- 1 Bock, K, Jensen, S R and Nielsen, B J (1976) Acta Chem Scand Ser B 30, 743
- 2 Bianco, A, Guiso, M, Iavarone, C, Passacantilli, P and Trogolo, C (1978) Gazz Chim Ital 108, 13
- 3 Iavarone, C, Sen, A, Trogolo, C and Villa, S (1983)

  Phytochemistry 22, 175
- 4 Inoue, K, Takeda, Y, Tanahashi, T and Inouye, H (1981) Chem Pharm. Bull 29, 970
- 5 Tietze, L-F (1973) Angew Chem 85, 763
- 6 Tietze, L-F (1974) Chem. Ber 107, 2499
- 7 Inouye, H, Saito, S, Taguchi, H and Endo, T (1969) Tetrahedron Letters 2347
- 8 Inouye, H, Takeda, Y, Saito, S, Nishimura, H and Sakuragi, R (1974) Yakugaku Zasshi 94, 577
- 9 Battersby, A R (1967) Pure Appl Chem 14, 117
- 10 Inoue, K, Takeda, Y, Tanahashi, T and Inouye, H (1981)

  Chem Pharm Bull 29, 981
- 11 Battersby, A R, Westcott, N D, Glüsenkamp, K-H and Tietze, L-F (1981) Chem Ber 114, 3439
- 12 Inouye, H, Tobita, S, Akiyama, Y, Ito, K and Shingu, T (1973) Chem Pharm Bull 21, 846